Sl.No.1306 Course Code: 71917103

## VINAYAKA MISSIONS RESEARCH FOUNDATION, SALEM

(Deemed to be University)

## **B.SC(CHEMISTRY) DEGREE EXAMINATION – November 2018**First Semester

## DSC - I - GENERAL CHEMISTRY- I

Time: Three hours Maximum: 70 marks

PART – A

 $(10 \times 2 = 20)$ 

### (Answer ALL Questions)

- 1. What are the fundamental particles of matter?
- 2. Write the Schrodinger's wave equation in one dimension
- 3. State Charles's law.
- 4. Define Collision diameter.
- 5. State Aufbau Principle.
- 6. What is Electron Affinity?
- 7. How Nickel will be identified?
- 8. What are the basic requirements of primary standard solution?
- 9. Write the canonical structures of CO<sub>2</sub>?
- 10. Write short note on Baeyer's Strain theory

 $\mathbf{PART} - \mathbf{B} \tag{4 x 5 = 20}$ 

## (Answer ALL Questions)

11. a) Describe the Rutherford's atom model.

(OR)

- b) Draw the diagram of d orbital.
- 12. a) Write any 3 postulates of Kinetic theory of gases

 $(\mathbf{OR})$ 

- b) Explain the important feature of molecular distribution curve.
- 13. a) State and explain Aufbau principle

(OR)

- b) Discuss about (i) p block elements (ii) f block elements
- 14. a) How cadmium is identified in the presence of copper?

(OR)

b) Describe Baker – Nathan effect with suitable example

PART - C

 $(3 \times 10 = 30)$ 

#### (Answer any THREE Questions)

- 16. Describe Sommerfeld's atom model. Give its drawbacks.
- 17. Discuss the Maxwell's distribution of molecular velocities
- 18. Discuss about Transition elements and Inner transition elements.
- 19. Explain the principle, advantages and apparatus used in Semi-micro analysis.
- 20. Explain Diels Alder reaction with suitable examples.

Sl.No.1934 Course Code: 71917203

## VINAYAKA MISSIONS RESEARCH FOUNDATION, SALEM

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## B.SC(CHEMISTRY) DEGREE EXAMINATION – November 2018 Second Semester

#### DSC II – GENERAL CHEMISTRY - II

Time: Three hours Maximum: 70 marks

PART – A

 $(10 \times 2 = 20)$ 

## (Answer ALL Questions)

- 1. What is Inert pair effect? Give an example
- 2. Write the difference between S and P orbitals
- 3. What is the position of Hydrogen in Periodic Table?
- 4. What are silanes? Give an example.
- 5. What is electrophile?
- 6. Define elimination reaction
- 7. State Markovnikov's rule. Give an example
- 8. Write the Friedel Crafts alkylation and acylation reaction
- 9. Define surface tension
- 10. Define mole fraction

PART – B

 $(4 \times 5 = 20)$ 

## (Answer ALL Questions)

11. a) Briefly explain the factors affecting polarization effect

(OR)

- b) Describe valance bond theory with neat diagrams
- 12. a) Explain the preparation, properties, structure and uses of LiH

(OR)

- b) Differentiate carbides and hydrides
- 13. a) Write a detail note on  $S_N^{-1}$  reaction

(OR)

- b) Explain about Cis Trans elimination reaction
- 14. a) Explain Huckel theory of aromaticity with example

(OR)

b) Write a note on liquid crystals

PART - C

 $(3 \times 10 = 30)$ 

#### (Answer any THREE Questions)

- 15. Define molecular orbital theory
- 16. Write the Preparation, properties and technical applications of carbide
- 17. Explain about  $E_1 \& E_2$  mechanism
- 18. Explain about aromaticity and resonance structure of benzene
- 19. Write note on a) Refractive index b) Specific refraction c) Molar refraction

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## VINAYAKA MISSIONS RESEARCH FOUNDATION, SALEM

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## B.SC(CHEMISTRY) DEGREE EXAMINATION – November 2018 Second Semester

#### **BIOCHEMISTRY**

Time: Three hours Maximum: 70 marks

PART – A

 $(10 \times 2 = 20)$ 

## (Answer ALL Questions)

- 1. Write a note on SI Units.
- 2. Give the differences between Intermolecular and intramolecular hydrogen bonding
- 3. What are Disaccharides? Give examples
- 4. Write the general classification of lipids.
- 5. What is zwitter ion?
- 6. Write a note on Denaturation of protein.
- 7. List the composition of DNA.
- 8. Write the features of endonucleases.
- 9. Draw the structure of Niacin and Thiamine.
- 10. Write the deficiency symptoms of (a) Sodium (b) phosphorous.

PART – B

 $(4 \times 5 = 20)$ 

## (Answer ALL Questions)

11. a) Write a note on (i) Special symbols of SI Units (ii) Prefixes for SI Units

(OR)

- b) Write a note on (i) sigma bond (ii) pi-bond
- 12. a) Explain the straight chain structure of Glucose

 $(\mathbf{OR})$ 

- b) Write down the Biological importance of lipids
- 13. a) Write the classification Protein in detail?

(OR)

- b) Write a note of Denaturation and Renaturation of protein.
- 14. a) Write a note on (i) Denaturation (ii) Renaturation of nucleic acid

(OR)

b) Explain the nutritional importance of Minerals.

PART - C

 $(3 \times 10 = 30)$ 

## (Answer any THREE Questions)

- 15. Explain Molecular orbital theory.
- 16. Write the properties of Monosaccharides
- 17. Write the structural organization Protein?
- 18. Write a detail notes on Watson and Crick model of DNA and its properties.
- 19. Explain in detail about the sources of different minerals.

S.No.1225 Sub.Code:71917302

# VINAYAKA MISSION'S RESEARCH FOUNDATION, SALEM (Deemed to be University)

## **B.SC (CHEMISTRY) DEGREE EXAMINATION – November 2018**

#### **Third Semester**

#### **DSC I - FOOD & NUTRITION**

Time: Three Hours Maximum: 70 marks

#### **SECTION - A**

## Answer All questions $(10 \times 2 = 20)$

- 1 Write short note on Macronutrients.
- 2 Write the different categories of food adulteration.
- 3 What is food contamination?
- 4 What is Gastric lavage?
- 5 List the preparative operations in food industry.
- 6 What is pasteurization?
- 7 Define Vitamins.
- 8 Write short note on Scurvy.
- 9 What is meant by trace mineral?
- 10 Write the functions of following minerals (a) Calcium (b) Potassium.

#### **SECTION - B**

#### **Answer the following**

(4 X 5 = 20)

11.a Briefly explain about Micronutrients.

OR

- .b Briefly explain about functions of Vitamin E and Vitamin K.
- 12.a Write shortly about food poisoning and its causes.

OR

- .b How will investigate food poisoning cases?
- 13.a Write short note on contaminants in food material.

OR

- .b Write briefly on food preservation.
- 14.a Write the functions and deficiency symptoms of Vitamin D.

OR

.b Write the functions and deficiency symptoms of Potassium.

#### **SECTION -C**

III. Answer ANY **THREE** of the following questions:

 $(3 \times 10 = 30)$ 

15 Write the functions of different minerals in detail.

- 16 Explain in detail about food poisoning.
- 17 Explain in detail on preparative operations in food industry.
- 18 Write in detail about (i) Vitamin B5 (ii) Vitamin C.
- 19 Explain in detail the sources, functions and deficiency symptoms of Sodium.

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